

Atomic defects and doping of monolayer NbSe₂

Nguyen, L.; Komsa, H.-P.; Khestanova, E.; Kashtiban, R.; Peters, J. P.; Lawlor¹, S.;
Sanchez, A. M.; Sloan, J.; Gorbachev, R.; Grigorieva, I.; Krasheninnikov, A. V.;
Haigh, S. J.;

Originally published:

February 2017

ACS Nano 11(2017), 2894-2904

DOI: <https://doi.org/10.1021/acsnano.6b08036>

Perma-Link to Publication Repository of HZDR:

<https://www.hzdr.de/publications/Publ-25386>

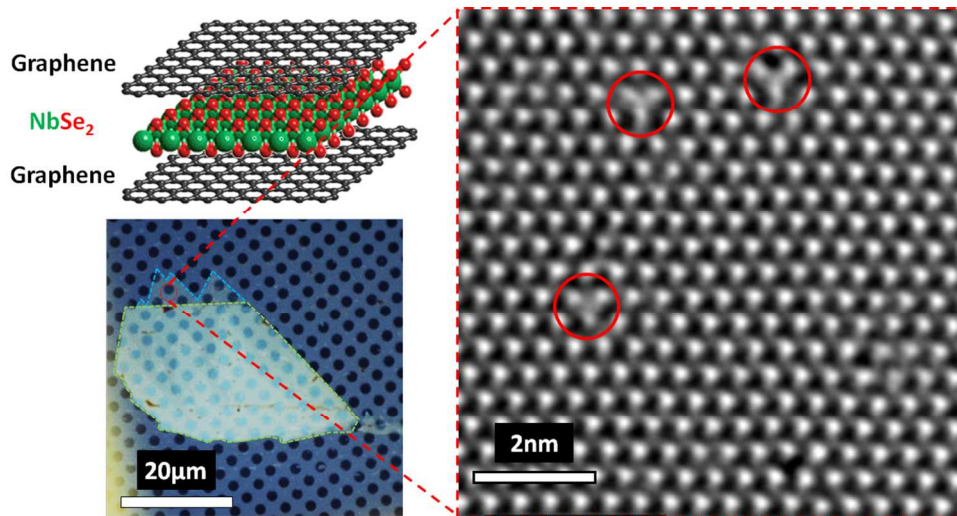
Release of the secondary publication
on the basis of the German Copyright Law § 38 Section 4.

This document is confidential and is proprietary to the American Chemical Society and its authors. Do not copy or disclose without written permission. If you have received this item in error, notify the sender and delete all copies.

Atomic Defects and Doping of Monolayer NbSe₂

Journal:	ACS Nano
Manuscript ID	nn-2016-080367.R3
Manuscript Type:	Article
Date Submitted by the Author:	13-Feb-2017
Complete List of Authors:	<p>Nguyen, Lan; University of Manchester, School of Materials Komsa, Hannu-Pekka; Aalto University, Department of Applied Physics Khestanova, Ekaterina; University of Manchester, School of Physics & Astronomy Kashtiban, Reza; The University of Warwick, Department of Physics Peters, Jonathan; The University of Warwick, Department of Physics Lawlor, Sean; University of Manchester, School of Materials Sanchez, Ana M.; University of Warwick, Department of Physics Sloan, Jeremy; University of Warwick, Department of Physics Gorbachev, Roman; University of Manchester, School of Physics & Astronomy Grigorieva, Irina; University of Manchester, The School of Physics and Astronomy Krasheninnikov, Arkady; Helmholtz-Zentrum Dresden-Rossendorf, Ion Beam Centre ; National University of Science and Technology MISiS; Aalto University, Department of Applied Physics Haigh, Sarah; University of Manchester, School of Materials</p>

SCHOLARONE™
Manuscripts



103x52mm (300 x 300 DPI)

Atomic Defects and Doping of Monolayer NbSe₂

Lan Nguyen¹, Hannu-Pekka Komsa², Ekaterina Khestanova³, Reza J. Kashtiban⁴, Jonathan J.P. Peters⁴, Sean Lawlor¹, Ana M. Sanchez⁴, Jeremy Sloan⁴, Roman V. Gorbachev³, Irina V. Grigorieva³, Arkady V. Krasheninnikov^{5,6,7}, Sarah J. Haigh^{1}*

Corresponding Author

*sarah.haigh@manchester.ac.uk

1. School of Materials, University of Manchester, Oxford Road, M13 9PL, United Kingdom
2. COMP Centre of Excellence, Department of Applied Physics, Aalto University, P.O. Box 11100, FI-00076 Aalto, Finland
3. School of Physics and Astronomy, University of Manchester, Oxford Road, M13 9PL, United Kingdom
4. Department of Physics, University of Warwick, Coventry, CV4 7AL, United Kingdom
5. Helmholtz-Zentrum Dresden-Rossendorf, Institute of Ion Beam Physics and Materials Research, 01328 Dresden, Germany
6. National University of Science and Technology MISiS, Leninskiy Prospekt, Moscow, 119049, Russian Federation
7. Department of Applied Physics, Aalto University, P.O. Box 11100, FI-00076 Aalto, Finland

1
2
3 **KEYWORDS:** Air-sensitive 2D Crystals, Graphene Encapsulation, Monolayer NbSe₂,
4
5 Transition Metal Dichalcogenides, Atomic Resolution TEM, Defect Dynamics, Pt Doping
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 **ABSTRACT:** We have investigated the structure of atomic defects within monolayer NbSe₂
4
5 encapsulated in graphene by combining atomic-resolution transmission electron microscope
6
7 (TEM) imaging, density functional theory (DFT) calculations and strain mapping using
8
9 geometric phase analysis (GPA). We demonstrate the presence of stable Nb and Se
10
11 monovacancies in monolayer material and reveal that Se monovacancies are the most
12
13 frequently observed defect, consistent with DFT calculations of their formation energy. We
14
15 reveal that adventitious impurities of C, N and O can substitute into the NbSe₂ lattice
16
17 stabilizing Se divacancies. We further observe evidence of Pt substitution into both Se and
18
19 Nb vacancy sites. This knowledge of the character and relative frequency of different atomic
20
21 defects provides the potential to better understand and control the unusual electronic and
22
23 magnetic properties of this exciting two-dimensional material.
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 Two-dimensional (2D) crystals such as graphene and the transition metal dichalcogenides
4 (TMDCs) have attracted considerable scientific interest in recent years due to their unique
5 properties and potential for revolutionary technological applications.¹⁻³ Aberration-corrected
6 transmission electron microscopy (TEM) techniques have played a crucial role in unravelling
7 the structure-property relationships of these materials at the atomic level.⁴⁻⁹ In particular,
8 TEM can reveal the presence of local deviations from the pristine atomic structure, such as
9 intrinsic defects, dislocations or dopant atoms, which can directly affect the properties of the
10 material.⁴⁻⁹

11
12 Recent work has reported the tailoring of the electronic and functional properties of 2D
13 crystals by deliberately introducing defects and/or doping the material with foreign atoms.^{5, 8,}
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

10-17 This offers promising pathways for engineering 2D crystals to a broad range of
applications in sensors and electronics.¹⁸⁻²⁰ In particular, the nanoscale engineering of
magnetic properties of 2D crystals *via* substitutional doping is a promising research area.^{5, 10,}
12, 13, 16, 17

35 Niobium diselenide (NbSe₂) is a 2D crystal which is gathering much attention; it
36 demonstrates superconductivity at low temperatures (<10K)²¹⁻²⁵, charge density wave
37 formation from ≤33K in bulk to ≤145K in the monolayer^{22, 26-29} and excellent
38 photoconductivity.³⁰ Like the more commonly studied TMDCs, NbSe₂ has a trigonal
39 prismatic crystal structure with strong covalent bonding of Se-Nb-Se atoms within the plane
40 and weak van-der Waals bonding between layers, allowing for mechanical^{21, 24, 25, 31, 32} and
41 liquid exfoliation³³ of ultrathin sheets. However, understanding and optimising the unique
42 properties of NbSe₂ has been held back by the severe degradation of the material, which
43 occurs under ambient conditions when exfoliated to a few layers.^{21, 31, 34, 35}

1
2
3 We have recently demonstrated superconductivity for an exfoliated monolayer material by
4
5 studying NbSe₂ monolayers.³¹ This was achieved by preparing samples using mechanical
6
7 exfoliation in a highly controlled inert glove box environment, coupled with encapsulation
8
9 using structurally stable 2D materials (graphene or hexagonal BN) and tunnelling contacts.
10
11 The encapsulation protects the NbSe₂ from the environment preserving superconducting
12
13 properties, whereas no transition to the superconducting phase was observed for monolayers
14
15 prepared under ambient conditions. Curiously, a number of properties change dramatically
16
17 from bilayer NbSe₂ to monolayer, such as a significant drop in the superconducting transition
18
19 temperature³¹ and the increase of temperature at which charge density waves (CDW)²⁶ are
20
21 observed. The origin of this behaviour remains unknown, but a likely hypothesis is that it is
22
23 related to atomic scale defects, the nature of which could be elucidated by structural studies.
24
25 Only by understanding the degradation of such materials at the atomic level can their
26
27 properties be fully realised and exploited.
28
29
30
31

32
33 A recent scanning tunnelling microscopy study has shown that defects in bulk NbSe₂ play a
34
35 crucial role in stabilising nanoscale regions of CDW order at temperatures higher than the
36
37 bulk CDW transition temperature.³⁶ These nanoscale regions of CDW ordering increase in
38
39 size as the temperature is lowered, until the CDW phase persists throughout the crystal.³⁶
40
41 Despite the importance of these defects, to our knowledge no-one has yet studied defects in
42
43 monolayer NbSe₂ material. TEM provides a complementary approach for atomic scale
44
45 imaging, and can be combined with spectroscopic imaging techniques to gain information on
46
47 local elemental distribution and electronic structure.^{37, 38} However to date, the difficulty of
48
49 preparing representative samples has limited electron microscopy studies of NbSe₂ to
50
51 relatively low resolution imaging and electron diffraction.^{39, 40} Here, using graphene
52
53 encapsulation to preserve the structure we provide atomic resolution TEM imaging of
54
55 monolayer NbSe₂. The presence of the graphene sheets preserves the lattice structure of the
56
57
58
59
60

1
2
3 material for more than 10 months in ambient conditions; in stark contrast to unprotected
4 samples which degrade after a few hours. However, when imaged at the atomic scale we find
5 the encapsulated NbSe₂ contains significant populations of atomic defects which are not the
6 simply the result of electron beam irradiation. We correlate our experimental imaging results
7 with extensive density functional theory (DFT) investigations of defect formation energies.
8
9
10
11
12

13 14 15 **Results/Discussion**

16
17 To prepare specimens for TEM investigations, thin NbSe₂ flakes were exfoliated in an inert
18 environment and encapsulated between two graphene sheets (Figure 1b) before being
19 transferred onto gold quantifoil grids.⁴¹ The thinnest regions of such flakes were investigated
20 by selected-area electron diffraction (SAED) in the TEM (Figure 1d). Our multislice
21 calculations⁴² show that by analysing the ratio of the first $\{\bar{1}010\}$ and second order $\{\bar{2}110\}$
22 diffraction spot intensities it is possible to identify monolayer regions of NbSe₂ similar to
23 previous electron diffraction experiments performed on 2D crystals.⁴³⁻⁴⁷ These simulations
24 show that the ratio of $\{\bar{1}010\}/\{\bar{2}110\}$ diffraction spot intensities has a value of 1.16 for
25 monolayer NbSe₂, 0.01 for bilayer, and 0.16 for trilayer. For thicker samples the values of
26 these ratios continue to decrease in an oscillatory fashion attenuating close to zero for even
27 numbers of layers as shown in supporting information Figure S1 (in addition to similar values
28 obtained from the $\{0\bar{1}10\}/\{\bar{2}110\}$). Our experimental results show a
29 $\{\bar{1}010\}/\{\bar{2}110\}$ intensity ratio of 0.56. This is much higher than would be obtained for
30 bilayer or thicker flakes indicating that the region captured by the 40 μm diameter SAED
31 aperture must contain largely monolayer material. That this is less than the 1.16 predicted for
32 perfect monolayer suggests that the $\sim 1\mu\text{m}$ diameter region sampled also contained some few-
33 layer NbSe₂. We have further confirmed the thinnest sample regions to be monolayer using
34 atomic resolution focal series TEM imaging and simulations (SI Figure S2).
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 Aberration corrected low voltage TEM imaging (80kV) revealed the atomic structure of our
4 encapsulated monolayer NbSe₂. Figure 2 shows examples of the different types of vacancy
5 defects that were observed experimentally, together with comparison to image simulations
6 produced from our DFT optimised atomic models for vacancies in this material. Such
7 analysis allows us to determine the nature of different types of atomic vacancies,
8 demonstrating a good agreement between experimental images and simulations for niobium
9 vacancies (V_{Nb}), selenium monovacancies (V_{Se}) and selenium divacancies, (V_{Se_2}), where the
10 two missing selenium atoms are in the same location when viewed normal to the flake along
11 $\langle 0001 \rangle$. Experimental images were low pass filtered to remove the small contrast
12 contributions from the graphene lattice and clearly visualise defects in the NbSe₂ lattice (SI
13 Figure S3). Qualitatively, the selenium vacancy defects form white triangles of contrast at
14 Scherzer defocus with a greater increase in intensity for the central selenium site found for
15 selenium divacancies compared to selenium monovacancies. The niobium vacancies appear
16 as triangles of contrast with the opposite orientation to the selenium vacancy features (an
17 increase of intensity on the Nb site compared to the pristine lattice).
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37

38 Our previous work has demonstrated that even though TMDCs with defects produced by the
39 electron beam are not in thermodynamic equilibrium, the relative abundance of defects and
40 atom displacement energies correlate with defect formation energies.^{8, 48} Here we have used
41 DFT to analyse the formation energies for the different vacancies we observe experimentally
42 in NbSe₂ (SI Table T1) (for details of modelling see 'Methods' section). At neutral
43 conditions (mid-point between the Se-rich or Nb-rich limits), formation energies for V_{Se} and
44 V_{Nb} monovacancies were found to be 1.05 eV and 3.49 eV respectively. These values are
45 lower than those found for more common TMDCs, *e.g.*, in MoS₂ under similar conditions,⁴⁹
46 which is in agreement with the consensus in the literature that NbSe₂ degrades more easily
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 than these other TMDCs. As well as monovacancies we also observe a significant number of
4
5 Se divacancies (the relative frequency of different defects is considered later). However, our
6
7 DFT calculations predict that V_{Se2} has a high formation energy of 3.53 eV. The presence of
8
9 V_{Se2} defects in the $NbSe_2$ sheet is therefore a puzzle; to reduce the total energy one would
10
11 expect the crystal to relax to form two V_{Se} defects with the lower combined formation energy
12
13 ($2V_{Se} = 2.09$ eV). Furthermore, our first principles calculations show that all vacancy
14
15 interactions in $NbSe_2$ are repulsive, which would drive apart pairs of Se monovacancies once
16
17 the divacancy had dissociated (see supplementary material Figure S4).
18
19

20
21 To explain the presence of significant V_{Se2} defects in our material, we have performed DFT
22
23 calculations investigating the stability of different elemental substitutions into the V_{Se2} sites.
24
25 Cu, Si, O, C, N, and H were investigated as these are common contaminants introduced
26
27 during the exfoliation and flake transfer process. Pt was also included, having been
28
29 deliberately evaporated on the surface during sample preparation. The heavier elements could
30
31 quickly be ruled out as TEM image simulations show that these produce a large reduction in
32
33 contrast (SI Figure S5). We then considered the dissociation energy associated with
34
35 transforming an impurity-substituted divacancy to an impurity-substituted vacancy and a bare
36
37 vacancy, *i.e.*, $E_{dis} = [E(X_{Se})+E(V_{Se})]-[E(X_{Se2})+E(pristine)]$, where X is the impurity atom
38
39 (Table 1). We also consider the case bare divacancy, V_{Se2} , transforming into two bare
40
41 monovacancies, $2V_{Se}$ (top row Table 1) which is not energetically favourable as discussed
42
43 above. Similarly, hydrogen substitution has a positive dissociation energy so is unable to
44
45 stabilize the divacancy. In contrast, C, N, and O impurities all have negative dissociation
46
47 energies, and thus should lead to stable impurity-substituted divacancies. The availability of
48
49 C, N and O atoms is assured in such a system from trapped hydrocarbon contamination^{41, 50}.
50
51 C atoms could also be displaced from the encapsulating graphene sheets during electron
52
53 beam irradiation^{51, 52}. Supporting information Figure S6 shows structural DFT models
54
55
56
57
58
59
60

1
2
3 illustrating how substitutional carbon atoms relax inwards towards the plane of Nb atoms
4
5 when substituting a Se divacancy. Unfortunately, it is not possible to experimentally verify
6
7 the presence of light substitutional impurities using our TEM image simulations and linescans
8
9 as illustrated in Figure 3g-h.
10

11
12
13
14 To determine whether the Se divacancies were indeed stabilised by adventitious impurities,
15
16 we employed geometric phase analysis (GPA)⁵³ to analyse the strain components surrounding
17
18 these defects (see methods and SI Figure S7 for details). Figure 3 compares vertical and
19
20 horizontal strain components from an experimental image of a typical divacancy type defect
21
22 with the strain predicted from analysis of HRTEM image simulations calculated using our
23
24 DFT relaxed defect structures. The e_{yx} and e_{yy} strain components show the key characteristics
25
26 of tension and compression produced in the lattice; only V_{Se2} , H_{Se2} , C_{Se2} , N_{Se2} , O_{Se2} , and V_{Nb}
27
28 defects are predicted to cause significant strain. Of these, C_{Se2} , N_{Se2} and O_{Se2} match well to
29
30 the experimental results (Figure 3e, SI Figure S8). This provides experimental evidence
31
32 supporting our earlier theoretical conclusion that such divacancy defects must be stabilised by
33
34 C, N or O impurities, and we have chosen to highlight the C-substituted case in Figure 3
35
36 since this is the most commonly available of the three impurities. Our DFT calculations
37
38 predict that oxygen substitution can also stabilise Se monovacancies but these do not produce
39
40 significant strain on the lattice or contrast for TEM imaging so experimentally it is difficult to
41
42 distinguish these from bare vacancies.
43
44
45
46
47
48

49 All TEM studies must consider the potential for electron beam induced defect creation. In
50
51 this work, all monolayer NbSe₂ areas studied were found to be highly defective even after
52
53 minimal electron dose (received while identifying the correct region of interest using low
54
55 magnification imaging). In order to study the potential for beam induced defect creation as
56
57
58
59
60

1
2
3 well as investigate the relative frequency of occurrence of different types of defects, we have
4
5 measured changes in the number of defects in response to prolonged electron imaging (Figure
6
7 4). The $\sim 175\text{nm}^2$ region of material analysed here contained 28 V_{Se} monovacancies at initial
8
9 observation (Figure 4a). V_{Se} was overwhelmingly the most commonly observed point defect;
10
11 in agreement with our DFT calculations which predict a lower formation energy for V_{Se}
12
13 compared to the other defects we observed (SI Table T1). After atomic resolution imaging
14
15 with a typical high electron dose rate ($1.4 \times 10^6 \text{e} \cdot \text{\AA}^{-2} \text{s}^{-1}$) for 46s the number of Se
16
17 monovacancies increases to 42 (Figure 4b). Subsequent imaging with a reduced dose rate
18
19 ($1 \times 10^3 \text{e} \cdot \text{\AA}^{-2} \text{s}^{-1}$) for 43s reduces the number of these defects to 29. This defect population then
20
21 remains stable even after extended 'high dose rate' ($6.49 \times 10^7 \text{e} \cdot \text{\AA}^{-2}$) exposure (Figure 4d).
22
23 Under the same imaging conditions, the number of isolated $V_{\text{Se}2}$ defects decreases from 11 to
24
25 6 and the V_{Nb} defects remains stable at 2). We also observe extended defect regions, the area
26
27 of which increases slightly throughout imaging from 26.7nm^2 to 32nm^2 (Figure 4a-d). Figures
28
29 4e and 4f summarise the observed changes in the numbers and concentrations of each type of
30
31 defect. The point defect concentration increases initially from 0.82 to 0.98% during high dose
32
33 rate imaging (from Figure 4a-b), then decreases to 0.70% after low dose rate imaging (Figure
34
35 4c) and subsequently is relatively stable. The large number of defects present at first
36
37 observation, together with the overall reduction in point defects during imaging demonstrates
38
39 that they are not the result of electron beam damage. We must therefore conclude either that a
40
41 high proportion of the defects we observe are intrinsic to the crystal, or that defects occur
42
43 despite the inert conditions used for mechanical exfoliation or that graphene encapsulation
44
45 only partially stabilises the material.
46
47
48
49
50

51
52
53
54 We now consider the possible mechanism behind our observed increase in the number of V_{Se}
55
56 point defects during initial high dose rate imaging. TMDCs are often found to be highly
57
58
59
60

1
2
3 susceptible to knock-on damage but our DFT calculations have demonstrated that at 80kV the
4
5 energy of the electron beam is well below the knock-on displacement threshold for both Nb
6
7 and Se in NbSe₂ (SI Figure S9), as both species have relatively high atomic masses.
8
9 Furthermore, we note that beam induced defect formation is likely to be suppressed by the
10
11 encapsulating graphene sheets, which can reduce charging and heating effects as observed
12
13 during (S)TEM imaging of MoS₂ monolayers.^{51,52} It is feasible that other mechanisms such
14
15 as chemical etching⁵⁴ or a combination of electronic excitations and knock-on damage may
16
17 facilitate the creation of beam induced defects even at accelerating voltages well below the
18
19 knock-on damage threshold.^{54, 55} Frame-by-frame analysis of TEM image series has
20
21 demonstrated that many defects are mobile during imaging (Figure 5 and SI Videos V1a-b).
22
23 This opens up the possibility that defects may migrate into the field of view.. Under low dose
24
25 rate conditions we observe not a slower increase (as might be expected) but an overall
26
27 reduction in the number of point defects. We explain this as being a result of the presence of
28
29 two competing effects: point defect generation and point defect annihilation (or coalescence
30
31 into larger defective regions). At a low dose rate, the rate of annihilation and/or coalescence
32
33 of defects must be higher than the rate of defect generation. Vacancy coalescence to form
34
35 larger extended defects would result in an increase in the total defective area during imaging
36
37 and this is indeed observed (Figure 4f).
38
39
40
41
42
43
44

45 To consider further the potential for coalescence of mobile point defects we have performed a
46
47 quantitative analysis of point defect clustering for Figure 4a-d in Figure 4g (see SI for details
48
49 of calculation). The distance between point defects and their next nearest neighbour defect
50
51 are considered with 'clusters' being those occupying nearest neighbour Se columns or second
52
53 nearest neighbour Se columns. We observe that on average over these images no clustering
54
55 occurs for ~40% of defects, ~45% of defects are spaced at second nearest neighbour sites and
56
57
58
59
60

1
2
3 the remainder ~15% are at next nearest sites. As imaging progressed there was an increasing
4
5 tendency for defects to be isolated which is in agreement with our DFT predictions for
6
7 repulsive interactions between defects (SI Figure S4). The most energetically favourable
8
9 configuration is for second nearest neighbour sites, in agreement with our experimental
10
11 results. This contrasts with the defect coalescence behaviour observed for other transition
12
13 metal dichalcogenides such as MoS₂ which preferably form agglomerated lines of adjacent
14
15 defects.⁵ Indeed, we observe that in NbSe₂ a line defect can be spontaneously generated but
16
17 that it is not stable and quickly transforms to a larger defect cluster (Figure 4d).
18
19

20
21
22 Closer analysis of the behaviour of the dynamics of individual defect clusters is presented in
23
24 Figure 5. Figure 5a shows the creation and subsequent annihilation of an isolated Nb vacancy
25
26 in a pristine region of the crystal. Such defects are rare (Figure 4e) and unstable, which is
27
28 consistent with their high formation energy predicted from DFT calculations (SI Table T1).
29
30 Figure 5b illustrates the appearance of a selenium monovacancy within the pristine lattice,
31
32 which then transforms to a divacancy. This may occur by the formation of an isolated V_{Se}
33
34 defect, which is then stabilised by a mobile contaminant atom such as C, N or O and then
35
36 further loss of the other Se atom in the site. However, we have already demonstrated that
37
38 divacancy defects are unstable and the same process can occur in reverse; as shown in Figure
39
40 5c for a defect in a different region of the crystal. These observations further support a
41
42 mechanism for the formation of Se divacancies as well as their healing which involves a
43
44 transition to an intermediate energy monovacancy configuration, representing much smaller
45
46 energy step than the transition straight from pristine to divacancy and vice versa. Figure 5d
47
48 shows the apparent dynamic movement of a selenium divacancy between adjacent sites
49
50 together with the appearance of a selenium monovacancy next to the divacancy. This further
51
52 illustrates the mobility and local interactions of these defects under the electron beam.
53
54
55
56
57
58
59
60

1
2
3
4
5 We calculated migration barrier for Se vacancy diffusion using nudged elastic band (NEB)
6 method with the results shown in Figure 6a. The calculated migration barriers for V_{Se}
7 diffusion is 0.89 eV and for V_{Se_2} dissociating to two neighbouring single vacancies 0.53 eV.
8 Assuming an attempt frequency of 10^{12} 1/s, these correspond to migration rates of 0.001 and
9
10 1300 s^{-1} at room temperature. Hence bare double vacancies should readily dissociate. During
11 the aging of several months, significant diffusion is expected. During the imaging, some of
12 the diffusion events may have been activated by the electron beam. The atomic structures
13 show that migrating Se atoms do not show a significant out of plane behaviour suggesting
14 that the encapsulating graphene sheets should not hinder their diffusion.
15
16
17
18
19
20
21
22
23
24
25
26
27

28
29 As vacancies are abundant in our samples, it is worth considering the electronic structure of a
30 single Se vacancy. The calculated band structure and density of states of the system with a Se
31 vacancy are shown in Figure 6b. The states from about -3 eV to -2 eV originate from the Nb-
32 d states. Similar to pristine NbSe_2 , the Fermi-level fills half of these states, which is
33 responsible for the CDW distortion at low temperatures (cf. SI Figure S10). Vacancies have a
34 small effect on the positions of these states, but results in several localized states in the band
35 gap above the Nb-d manifold, which could make them resolvable with STM. The partial
36 charge densities show that the states originate from a combination of the d-states at the
37 neighbouring Nb atoms.
38
39
40
41
42
43
44
45
46
47
48
49
50

51
52 In an attempt to determine the effectiveness of the graphene encapsulation for protecting the
53 NbSe_2 , we used energy dispersive X-ray (EDX) spectrum imaging to analyse elemental
54 segregation in the sample after 4 months of ambient storage. Figure 7b-d shows a region of a
55
56
57
58
59
60

1
2
3 graphene encapsulated NbSe₂ flake (without Pt) which contains extended strips associated
4
5 with a high oxygen content. High resolution imaging and diffraction of such regions reveals
6
7 that in these regions the NbSe₂ has transformed to amorphous Nb₂O₅ (See SI Figure S11).
8
9 Such degradation products have been previously identified by Raman spectroscopy
10
11 techniques,²¹ but not observed at this spatial resolution. Analysis of the orientation of the
12
13 oxidation strips showed no relationship to the crystal structure of the original NbSe₂ but
14
15 instead the direction corresponded to the one of the encapsulating graphene planes (SI Figure
16
17 S12). Graphene provides an impermeable barrier to gases⁵⁶⁻⁵⁸ but only when it makes a
18
19 perfect seal to the surface. The morphology of the oxidation suggests that there was a crease
20
21 in the sheet so that the graphene did not form a perfect atomically sharp interface and was
22
23 therefore unable to protect the NbSe₂ from degradation in this region We also reveal an
24
25 enhancement of oxidation at the step edges of encapsulated flakes (Figure 7). This agrees
26
27 with DFT calculations for other 2D flakes which have shown that edges are often susceptible
28
29 to oxidation.^{35, 59}
30
31
32
33
34
35

36 Finally, we consider the substitution of Pt into the NbSe₂ lattice as a route to tuning the
37
38 electronic and magnetic properties of the material. A 0.5nm layer of Pt was originally
39
40 evaporated on the surface of the encapsulated NbSe₂ with the dual aim of producing
41
42 substitutional defects as well as catalytically reducing surface contamination.⁶⁰⁻⁶² Figure 8a-
43
44 b) compares an image simulation of Pt substitution into a Se site shown alongside an atomic
45
46 resolution TEM image of a typical 'dark spot' defect feature found frequently while imaging.
47
48 However, the distinction between this defect and a Nb atom on a Se site, or other heavy
49
50 element substitution is difficult to make from high resolution image contrast alone (see SI
51
52 Figure S5). To solve this problem, we have employed high angle annular dark field
53
54 (HAADF) STEM imaging, where the contrast is highly sensitive to atomic number. This has
55
56
57
58
59
60

1
2
3 clear advantages when studying heavy atoms on a lighter substrate^{55, 63} and already proved
4 highly complementary to TEM imaging many other 2D materials.^{7, 64} Figure 8c-d) shows a
5 HAADF simulation and experimental image of a monolayer region, where the direct
6 relationship between intensity and atomic number allows unambiguous identification of the
7 location of Se columns and Nb sites as well as identification of individual Pt atoms in the
8 image. The location of the individual Pt atoms was found to be collocated on either Se or Nb
9 sites within the NbSe₂ lattice. Our DFT calculations suggest that Pt substitutions are more
10 energetically favourable than Pt adatoms (SI Table T1), which together with our HAADF-
11 STEM results suggests that Pt has substituted into both Nb and Se sites in the lattice.
12
13
14
15
16
17
18
19
20
21
22
23

24
25 The presence of Pt substituted in the NbSe₂ lattice might at first seem surprising, as the
26 samples were produced by evaporated Pt on the surface of the encapsulating graphene.
27 However, previous studies have shown metal-mediated etching of graphene can occur under
28 the action of the electron beam.^{55, 63, 65-67} This etching of the graphene sheet was visible
29 during TEM imaging (SI Videos V2a-e) and would allow Pt atoms to interact and substitute
30 into the underlying NbSe₂ lattice. Furthermore, our calculations indicate that substitutional
31 impurities in either Nb or Se sites can lead to pronounced stabilization of magnetic moment
32 patterns also known as spin density waves (shown in Figure S13). However, since the
33 calculations correspond to 0 K, we cannot assess the stability of these patterns at finite
34 temperatures.
35
36
37
38
39
40
41
42
43
44
45
46
47
48

49 **Conclusions**

50 In conclusion, we have observed a wealth of atomic scale defects in graphene encapsulated
51 monolayer NbSe₂. We find a high concentration of Se vacancies and divacancies, with the
52 latter being found to be passivated and stabilised with substitutional impurity atoms (C, O or
53
54
55
56
57
58
59
60

1
2
3 N). The monolayer material is found to have a high defect density even after a minimal
4
5 electron dose, and we further observed that the vacancy defect concentration was relatively
6
7 stable over prolonged electron beam imaging at 80 kV. This suggests that a high proportion
8
9 of the defects we observe are intrinsic to the crystal or that defects occur despite the inert
10
11 conditions used for mechanical exfoliation. Graphene encapsulation is found to be essential
12
13 to preserve the crystal structure of monolayer / few-layer material over prolonged exposure to
14
15 ambient conditions although creases in the sheets should be avoided to prevent ingress of
16
17 oxidising species. Substitution of Pt into both Nb and Se sites is found to occur readily
18
19 opening up a potential pathway for tailoring the intriguing electronic and magnetic properties
20
21 of NbSe₂.
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

Figures

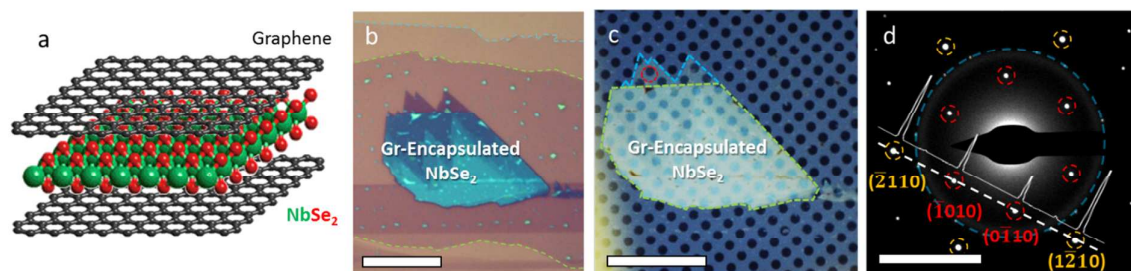


Figure 1: Preparation of graphene encapsulated NbSe₂ for TEM imaging by mechanical exfoliation in argon-environment glovebox. (a) Schematic of graphene encapsulated NbSe₂ stack. (b) Optical image showing NbSe₂ flake encapsulated between graphene sheets on oxidised silicon wafer (scale bar 20 μm). Thin regions were identified by the purple optical contrast. Edges of top and bottom encapsulating graphene sheets are highlighted by green and blue dashed lines. (c) Same Gr-NbSe₂-Gr stack transferred to a quantifoil TEM grid. Blue dashed line highlights thinnest NbSe₂ region (scale bar: 20μm). (d) SAED pattern from area of red circle in (c), the large blue dashed ring highlights position of graphene ($\bar{1}010$) reflections ($d=0.213\text{nm}$); small red circles highlight NbSe₂ ($\bar{1}010$) reflections ($d=0.296\text{nm}$); while small orange circles highlight NbSe₂, ($\bar{2}110$), reflections ($d=0.173\text{nm}$). Line profile highlights relative intensities of ($\bar{1}010$) and ($\bar{2}110$) diffraction spots used to determine specimen thickness (scale bar: 5nm^{-1}).

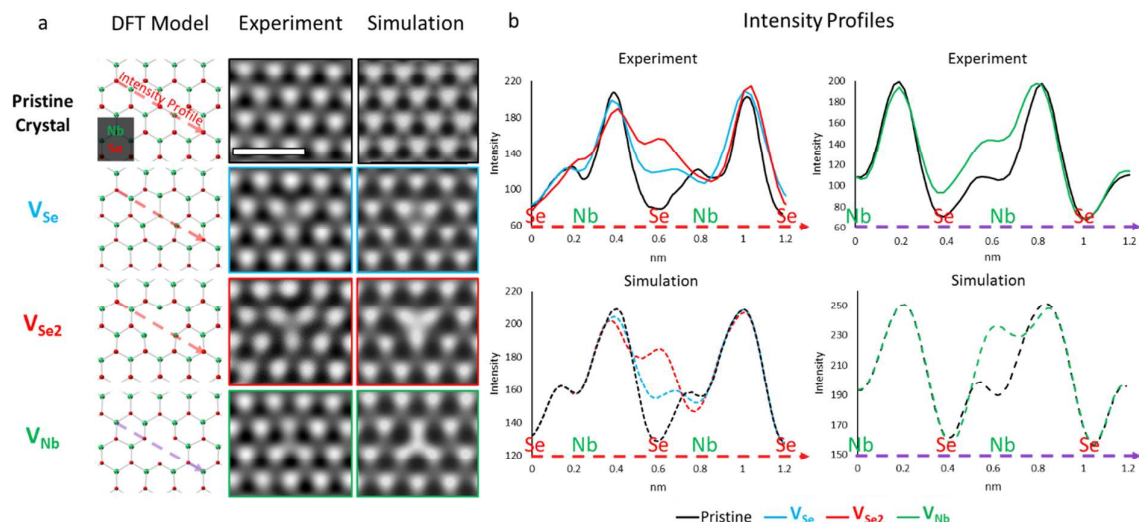


Figure 2: Experimental and simulated images for vacancy point defects in monolayer $NbSe_2$. (a) DFT relaxed structural models (left), high-resolution TEM images (centre) and multislice calculations (right) for pristine material (black border), a Se monovacancy (V_{Se} , blue border), Se divacancy (V_{Se2} , red border) and a Nb vacancy (V_{Nb} , green border) (scale bar: 1 nm). (b) Intensity profiles extracted along the armchair direction as shown by arrows in (a) for experiment (upper panels) and simulation (lower panels) demonstrating the ability to distinguish pristine crystal from V_{Se} , V_{Se2} (left) and V_{Nb} vacancies (right) using their individual intensity signatures.

Table 1: Energy change from dissociating a Se divacancy substituted with an impurity X into an impurity substituted monovacancy and a bare monovacancy. The top row V represents a divacancy V_{Se2} dissociating to two monovacancies. Other elements have their usual meanings of hydrogen, carbon, nitrogen, oxygen.

Element X	$X_{Se2} \rightarrow (X_{Se} + V_{Se})$
bare (V)	-1.44
H	-0.50
C	0.96
N	1.79
O	1.23

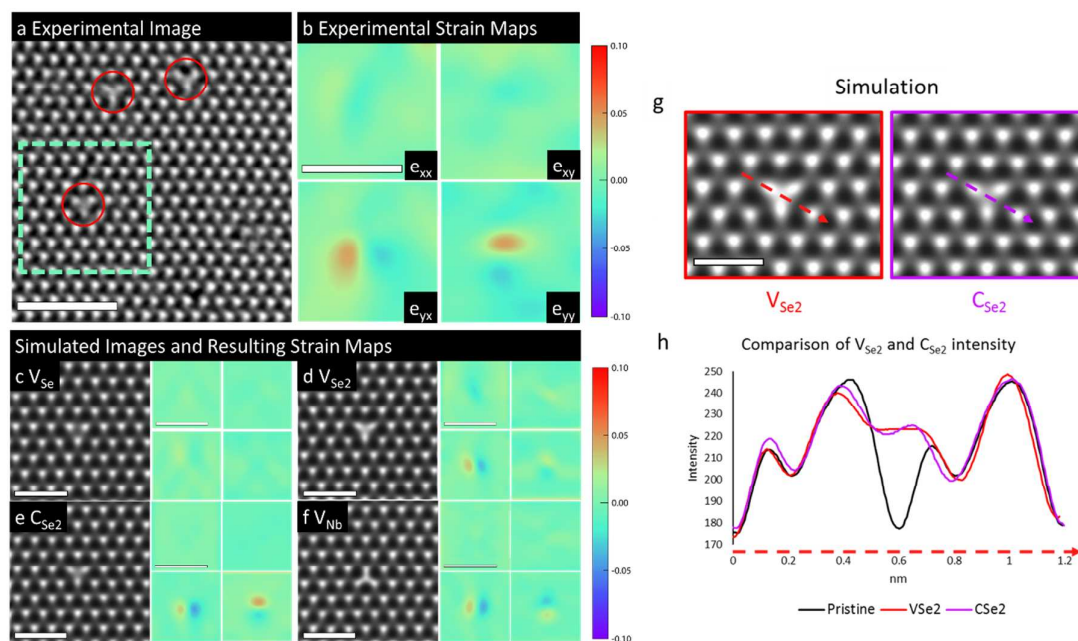


Figure 3: Geometric phase analysis (GPA) allows identification of impurity substitution into Se divacancy defects. (a) TEM image of a region containing three divacancy type defects (circled red) and (b) corresponding strain component maps for representative defect within green box in a). In e_{yx} and e_{yy} regions of tension and compression are observed horizontally and vertically across the defect (scale bars: 2nm). (c-f) GPA strain analysis from TEM image simulations for: (c) a Se monovacancy, V_{Se} ; (d) a Se divacancy, V_{Se2} ; (e) a carbon substituted Se divacancy, C_{Se2} ; and (f) a Nb monovacancy, V_{Nb} . The experimentally observed strain around the defect very closely matches the simulations for a C_{Se2} . (g) TEM image simulations comparing V_{Se2} and C_{Se2} . (h) Intensity profiles from the lines indicated on (g) reveal that TEM image contrast cannot be used to distinguish between these two types of defect. (Scale bars are 1nm for multislice images and 2nm for GPA maps).

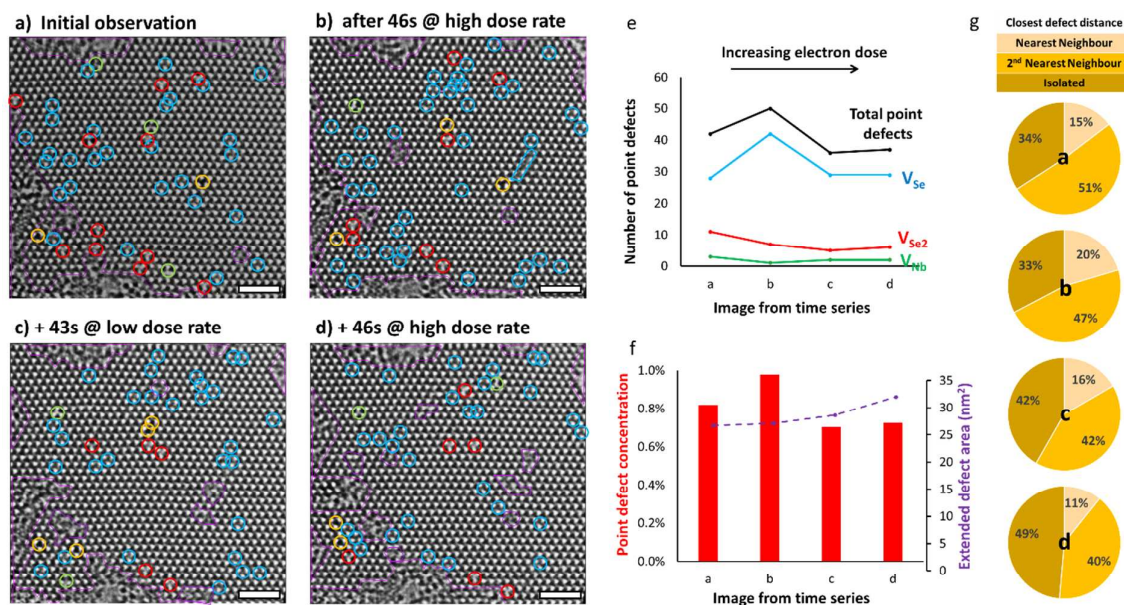


Figure 4: Point defect populations in NbSe₂ with continued electron irradiation. Atomic resolution TEM imaging of monolayer NbSe₂ (a) on first observation and (b) after continuous imaging with a high dose rate: $1.4 \times 10^6 \text{ e.}\text{\AA}^{-2}\text{s}^{-1}$ for 46s (total dose: $6.5 \times 10^7 \text{ e.}\text{\AA}^{-2}$). Immediately after acquiring (b) the beam was spread to reduce the electron dose rate to $\sim 1 \times 10^3 \text{ e.}\text{\AA}^{-2} \text{ s}^{-1}$. (c) the same area after 43s of low dose rate imaging (total accumulated dose of $6.5 \times 10^7 \text{ e.}\text{\AA}^{-2}$) and (d) shows the same area after another 46s of high dose rate imaging (total accumulated dose of $1.3 \times 10^8 \text{ e.}\text{\AA}^{-2}$). Coloured rings indicate the nature of point defects: Blue: Se monovacancy, V_{Se} ; Red: Se divacancy, V_{Se2} ; Green: Nb monovacancy, V_{Nb} and Yellow: Pt atom. Purple borders indicate larger, agglomerated regions of defects. Scale bars: 2nm. (e) Summarises the total number of point defects and the total area taken up by larger defect regions in the images (a-d). (f) Plots the percentage of point defects observed in (a-d). (g) Clustering analysis considering the distance between each point defect and its nearest neighbour defect. Any defects separated from their nearest neighbours by more than two nearest neighbour distances are considered to be ‘not clustered’ (isolated) (See SI Figure S4).

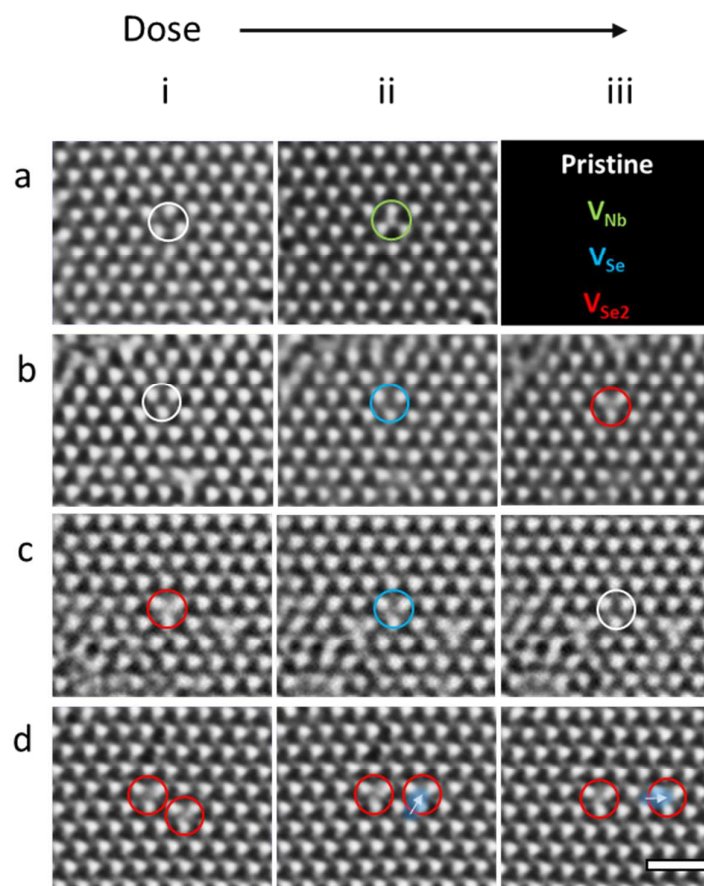


Figure 5: Time series imaging of point defect dynamics in monolayer NbSe₂. (a) Formation of an isolated Nb monovacancy (V_{Nb}) in the pristine lattice (at electron doses of $1.3 \times 10^7 \text{ e} \cdot \text{\AA}^{-2}$ and $1.6 \times 10^7 \text{ e} \cdot \text{\AA}^{-2}$ for (ai) and (aii) respectively). (b) Formation of an isolated V_{Se} monovacancy defect which subsequently transforms to a V_{Se2} divacancy type defect (doses of $3.2 \times 10^6 \text{ e} \cdot \text{\AA}^{-2}$, $9.7 \times 10^6 \text{ e} \cdot \text{\AA}^{-2}$, and $2.9 \times 10^7 \text{ e} \cdot \text{\AA}^{-2}$ for (bi), (bii) and (biii) respectively). (c) A V_{Se2} divacancy type defect which heals to form a V_{Se} monovacancy then to pristine lattice ((at electron doses of $2.0 \times 10^8 \text{ e} \cdot \text{\AA}^{-2}$, $2.27 \times 10^8 \text{ e} \cdot \text{\AA}^{-2}$ and $2.28 \times 10^8 \text{ e} \cdot \text{\AA}^{-2}$ for (ci), (cii) and (ciii) respectively). (d) A pair of V_{Se2} divacancies where one hops to adjacent lattice sites. (at electron doses of $1.4 \times 10^8 \text{ e} \cdot \text{\AA}^{-2}$, $1.7 \times 10^8 \text{ e} \cdot \text{\AA}^{-2}$ and $1.9 \times 10^8 \text{ e} \cdot \text{\AA}^{-2}$ for (di), (dii) and (diii) respectively) (scale bar: 1nm).

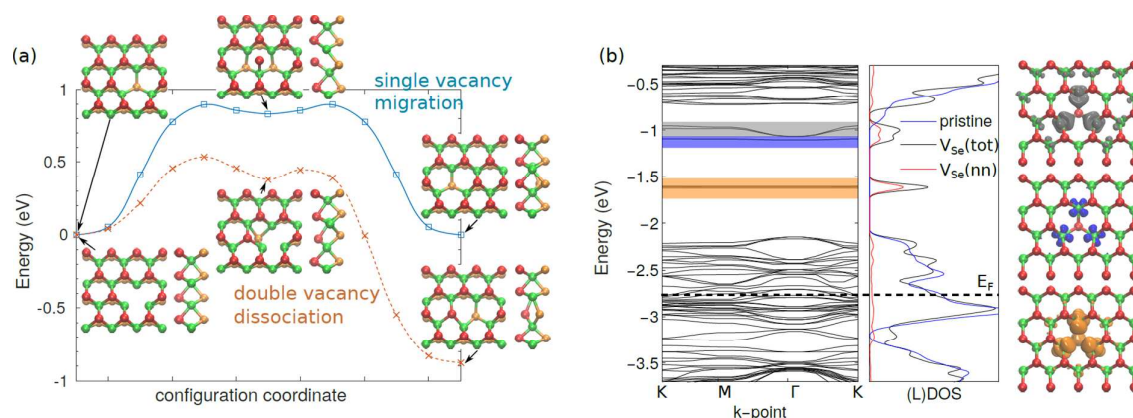


Figure 6: DFT calculations for vacancy migration and their effect on electronic structure. (a) Migration barriers for single vacancy diffusion (squares, blue) and divacancy dissociation to two neighbouring single vacancies (crosses, orange). The lines are guide to the eye. The initial, mid-point, and final atomic structures are also illustrated. (b) Band structure and density of states of the system with V_{Se} . Local DOS from the nearest neighbour Nb atoms to the vacancy and the DOS of pristine NbSe₂ are also shown. The localized defect states are visualized on the right as coloured in the band structure.

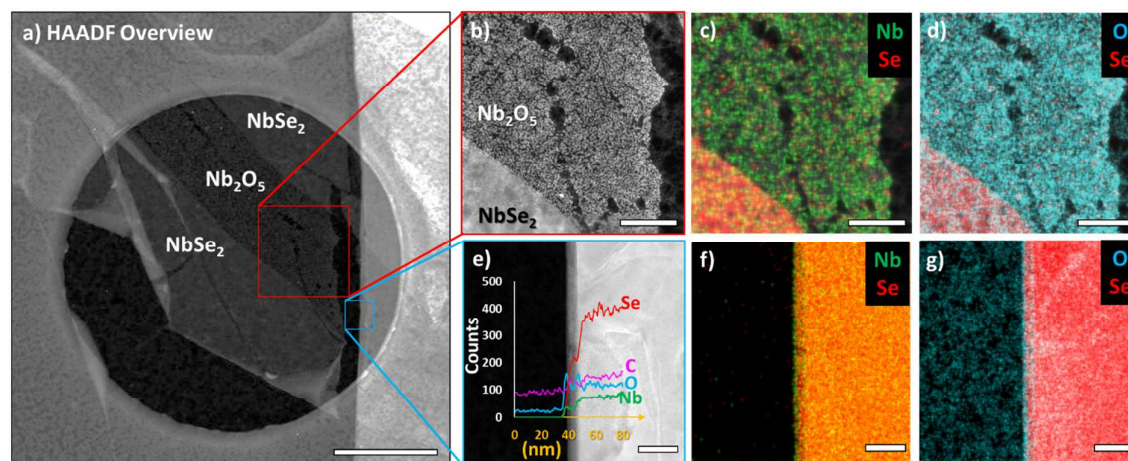


Figure 7: EDX Spectrum imaging revealing regions of oxidation for graphene encapsulated NbSe₂ crystals aged for 4 months in ambient conditions. a) HAADF overview of region of interest (scale bar: 500nm) b) HAADF, c) Nb+Se and d) O+Se elemental maps showing that a strip of crystal with a width of ~90 nm is oxidised to form Nb₂O₅ (Scale bar 100nm, see SI for EDX spectrum). e) HAADF, f) Nb+Se and g) O+Se elemental maps showing oxidation of a step flake edge. The enhancement of oxygen is further highlighted by the elemental line profile superimposed on e) (Scale bar: 20nm).

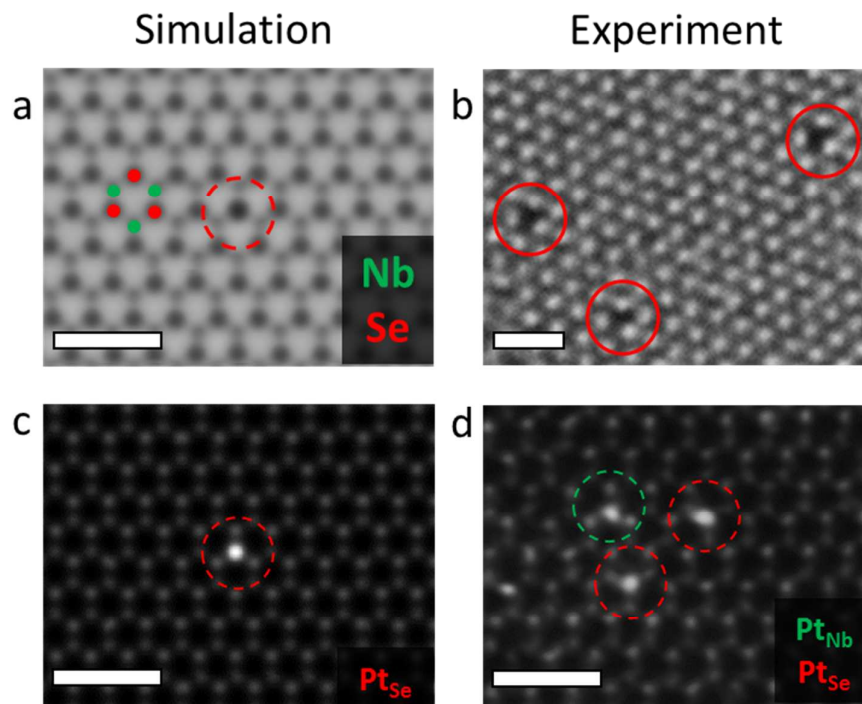


Figure 8: Observation of Pt substitution a) HRTEM image simulation of Pt substitution in a Se site. b) HRTEM experimental image of Pt substitutions into the lattice. c) HAADF STEM simulation of Pt substitution into a Se site, (Pt_{Se}). d) HAADF STEM experimental image of Pt substitutions into both Nb (Pt_{Nb}) and Se (Pt_{Se}) sites. Contrast clearly distinguishes Se columns from Nb and shows the presence of individual Pt atoms associated with both Se and Nb sites in the NbSe_2 lattice. All scale bars: 1 nm.

Methods/ Experimental:

Encapsulated NbSe₂ few layer samples were prepared by mechanical exfoliation in an argon-environment glovebox (levels of H₂O and O₂ below 0.1ppm).³¹ Crystals are transferred by a motorised micromanipulation station, which enables micrometre-precision encapsulation within graphene membranes.^{31, 41} The exfoliated crystals are typically <10 layers thick and have lateral dimensions up to a few microns. In addition to passivating the crystals, graphene encapsulation also reduces electron beam damage *via* charging or heating effects.^{51, 52} To minimise surface contamination, a thin 0.5nm layer of Pt is deposited on the top and bottom of the sample, which can catalytically remove residues when annealed.⁶⁰

Atomic-resolution imaging was performed on an aberration corrected JEOL ARM 200F TEM at 80kV at a constant dose rate of $1.41 \times 10^6 \text{ e.A}^{-2}\text{s}^{-1}$. High-resolution TEM and electron diffraction multi-slice simulations were performed using the JEMS package⁶⁸ (accelerating voltage: 80kV; C_c: 1.2mm; C_s: -0.005mm; C₅: 10mm; defocus: 5nm; defocus spread: 5nm and energy spread: 0.81eV). Atomic resolution STEM HAADF imaging of the encapsulated NbSe₂ was performed with a FEI Titan 80-200 ChemiSTEM with probe-side aberration correction at 200 kV and X-FEG electron source. STEM experiments were performed with a 21 mrad probe convergence angle and 80 pA probe current. The HAADF detector was collecting electrons scattered between 48 mrad and 191 mrad. HAADF simulations were performed with the QSTEM package.⁶⁹ Image processing was performed in Gatan Digital Micrograph software and GPA strain analysis was performed using the in house developed Strain++ program (for further information see SI).

DFT calculations were performed within the projector augmented wave framework, accounting for spin-polarization in the plane-wave basis as implemented in the code VASP.⁷⁰ ⁷¹ For defect calculations, we used the PBE exchange-correlation functional.⁷² 400 eV cutoff for the plane-wave basis yielded converged total energies. Within PBE, the lattice constant for the unreconstructed primitive cell is 3.49 Å. In the case of formation energy calculations, the defects were modeled using the 8x8 supercell and 3x3 k-point mesh. The migration barrier calculations were carried out using a smaller 6x6 supercell and 4x4 k-point mesh.

Author Contributions

The manuscript was written through contributions of all authors. All authors have given approval to the final version of the manuscript. LN and SJH performed diffraction, image simulations, HAADF STEM and EDX experiments, analysed the results and wrote the manuscript. SL helped with the analysis of results. HPK and AK performed DFT calculations and assisted with the discussion of results. EK, RG and IG prepared the samples and assisted with the discussion of the manuscript. RK performed HRTEM imaging of the specimen. JJPP, AMS and JS performed GPA calculations and assisted with discussion of the manuscript.

Supporting Information Available: Additional experimental details including: Electron diffraction and multislice calculation thickness determination, image processing details, DFT of clustering configurations, image simulations of substitutional defects, calculation details on the stability of impurity stabilised Se vacancies, GPA method details, additional GPA defect calculations, displacement cross sections for Se atoms, defect density calculation details, DFT of electronic structure of defects, EDX spectra of Nb₂O₅, diffraction revealing alignment of degraded NbSe₂ to crystallographic orientations of graphene, DFT of magnetic properties, DFT of the interaction of graphene encapsulation with NbSe₂, LDOS of a Se monovacancy, DFT of atomic CDW and SDW structures and table of defect formation energies.

Supplementary videos 1a-b: HRTEM of the dynamics of point defects.

Supplementary videos 2a-e: HAADF STEM of the dynamic etching of NbSe₂ with Pt under the electron beam.

This material is available free of charge *via* the Internet at <http://pubs.acs.org>.

Acknowledgement: The authors would like to thank the Engineering and Physical Sciences (EPSRC) U.K. Grants EP/G035954/1, EP/K016946/1, EP/M010619/1 and EP/J021172/1 and

1
2
3 Defence Threat Reduction Agency Grant HDTRA1-12-1-0013 for funding. Access to
4
5 aberration corrected electron microscopy was provided through the EPSRC SuperSTEM
6
7 laboratory (NS/A000016/1). AVK and HPK acknowledge the financial support from the U.S.
8
9 Army RDECOM *via* contract No. W911NF-15-1-0606 and from the Academy of Finland
10
11 (Project No. 286279) and through its Centres of Excellence Programme (Project No.
12
13 251748). AVK further thanks the Ministry of Education and Science of the Russian
14
15 Federation for the support in the framework of Increase Competitiveness Program of NUST
16
17 “MISiS” (#K2-2015-033).
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

References

1. Geim, A. K. Graphene: Status and Prospects. *Science* 2009, 324, 1530-1534.
2. Novoselov, K. S.; Fal'ko, V. I.; Colombo, L.; Gellert, P. R.; Schwab, M. G.; Kim, K. A Roadmap for Graphene. *Nature* 2012, 490, 192-200.
3. Chhowalla, M.; Shin, H. S.; Eda, G.; Li, L. J.; Loh, K. P.; Zhang, H. The Chemistry of Two-Dimensional Layered Transition Metal Dichalcogenide Nanosheets. *Nat. Chem.* 2013, 5, 263-275.
4. Lehtinen, O.; Komsa, H. P.; Pulkin, A.; Whitwick, M. B.; Chen, M. W.; Lehnert, T.; Mohn, M. J.; Yazyev, O. V.; Kis, A.; Kaiser, U.; Krasheninnikov, A. V. Atomic Scale Microstructure and Properties of Se-Deficient Two-Dimensional MoSe₂. *ACS Nano* 2015, 9, 3274-3283.
5. Komsa, H. P.; Kurasch, S.; Lehtinen, O.; Kaiser, U.; Krasheninnikov, A. V. From Point to Extended Defects in Two-Dimensional MoS₂: Evolution of Atomic Structure Under Electron Irradiation. *Phys. Rev. B* 2013, 88.
6. Robertson, A. W.; Warner, J. H. Atomic Resolution Imaging of Graphene by Transmission Electron Microscopy. *Nanoscale* 2013, 5, 4079-4093.
7. Lin, Y. C.; Dumcenccon, D. O.; Huang, Y. S.; Suenaga, K. Atomic Mechanism of the Semiconducting-to-Metallic Phase Transition in Single-Layered MoS₂. *Nat. Nanotechnol.* 2014, 9, 391-396.
8. Komsa, H. P.; Kotakoski, J.; Kurasch, S.; Lehtinen, O.; Kaiser, U.; Krasheninnikov, A. V. Two-Dimensional Transition Metal Dichalcogenides under Electron Irradiation: Defect Production and Doping. *Phys. Rev. Lett.* 2012, 109.
9. Kashtiban, R. J.; Dyson, M. A.; Nair, R. R.; Zan, R.; Wong, S. L.; Ramasse, Q.; Geim, A. K.; Bangert, U.; Sloan, J. Atomically Resolved Imaging of Highly Ordered Alternating Fluorinated Graphene. *Nat. Commun.* 2014, 5, 4902.
10. Lin, Y. C.; Dumcenco, D. O.; Komsa, H. P.; Niimi, Y.; Krasheninnikov, A. V.; Huang, Y. S.; Suenaga, K. Properties of Individual Dopant Atoms in Single-Layer MoS₂ : Atomic Structure, Migration, and Enhanced Reactivity. *Adv. Mater.* 2014, 26, 2857-2861.
11. Bangert, U.; Pierce, W.; Kepaptsoglou, D. M.; Ramasse, Q.; Zan, R.; Gass, M. H.; Van den Berg, J. A.; Boothroyd, C. B.; Amani, J.; Hofsass, H. Ion Implantation of Graphene - Toward IC Compatible Technologies. *Nano Lett.* 2013, 13, 4902-4907.
12. Tedstone, A. A.; Lewis, D. J.; O'Brien, P. Synthesis, Properties, and Applications of Transition Metal-Doped Layered Transition Metal Dichalcogenides. *Chem. Mater.* 2016, 28, 1965-1974.
13. Zhao, X.; Chen, P.; Xia, C. X.; Wang, T. X.; Dai, X. Q. Electronic and Magnetic Properties of n-type and p-doped MoS₂ Monolayers. *RSC Adv.* 2016, 6, 16772-16778.
14. Wang, H. T.; Wang, Q. X.; Cheng, Y. C.; Li, K.; Yao, Y. B.; Zhang, Q.; Dong, C. Z.; Wang, P.; Schwingschlogl, U.; Yang, W.; Zhang, X. X. Doping Monolayer Graphene with Single Atom Substitutions. *Nano Lett.* 2012, 12, 141-144.
15. Zhong, L.; Bruno, R. C.; Ethan, K.; Ruitao, L.; Rahul, R.; Humberto, T.; Marcos, A. P.; Mauricio, T. Defect Engineering of Two-Dimensional Transition Metal Dichalcogenides. *2D Mater.* 2016, 3, 022002.
16. Haldar, S.; Vovusha, H.; Yadav, M. K.; Eriksson, O.; Sanyal, B. Systematic Study of Structural, Electronic, and Optical Properties of Atomic-Scale Defects in the Two-Dimensional Transition Metal Dichalcogenides MX₂ M = Mo, W; X = S, Se, Te). *Phys. Rev. B* 2015, 92, 235408.
17. González-Herrero, H.; Gómez-Rodríguez, J. M.; Mallet, P.; Moaied, M.; Palacios, J. J.; Salgado, C.; Ugeda, M. M.; Veuillen, J.-Y.; Yndurain, F.; Brihuega, I. Atomic-Scale Control of Graphene Magnetism by Using Hydrogen Atoms. *Science* 2016, 352, 437-441.
18. Han, S. W.; Hwang, Y. H.; Kim, S.-H.; Yun, W. S.; Lee, J. D.; Park, M. G.; Ryu, S.; Park, J. S.; Yoo, D.-H.; Yoon, S.-P.; Hong, S. C.; Kim, K. S.; Park, Y. S. Controlling Ferromagnetic Easy Axis in a Layered MoS₂ Single Crystal. *Phys. Rev. Lett.* 2013, 110, 247201.
19. Tongay, S.; Suh, J.; Ataca, C.; Fan, W.; Luce, A.; Kang, J. S.; Liu, J.; Ko, C.; Raghunathanan, R.; Zhou, J.; Ogletree, F.; Li, J.; Grossman, J. C.; Wu, J. Defects Activated Photoluminescence in Two-

- 1
2
3 Dimensional Semiconductors: Interplay Between Bound, Charged, and Free Excitons. *Sci. Rep.* 2013,
4 3, 2657.
- 5 20. Li, H.; Tsai, C.; Koh, A. L.; Cai, L.; Contryman, A. W.; Fragapane, A. H.; Zhao, J.; Han, H. S.;
6 Manoharan, H. C.; Abild-Pedersen, F.; Norskov, J. K.; Zheng, X. Activating and Optimizing MoS₂ Basal
7 Planes for Hydrogen Evolution Through the Formation of Strained Sulphur Vacancies. *Nat. Mater.*
8 2016, 15, 48-53.
- 9 21. El-Bana, M. S.; Wolverson, D.; Russo, S.; Balakrishnan, G.; Paul, D. M.; Bending, S. J.
10 Superconductivity in Two-Dimensional NbSe₂ Field Effect Transistors. *Supercond. Sci. Technol.* 2013,
11 26.
- 12 22. Ugeda, M. M.; Bradley, A. J.; Zhang, Y.; Onishi, S.; Chen, Y.; Ruan, W.; Ojeda-Aristizabal, C.;
13 Ryu, H.; Edmonds, M. T.; Tsai, H. Z.; Riss, A.; Mo, S. K.; Lee, D. H.; Zettl, A.; Hussain, Z.; Shen, Z. X.;
14 Crommie, M. F. Characterization of Collective Ground States in Single-Layer NbSe₂. *Nat. Phys.* 2016,
15 12, 92-97.
- 16 23. Xi, X. X.; Wang, Z. F.; Zhao, W. W.; Park, J. H.; Law, K. T.; Berger, H.; Forro, L.; Shan, J.; Mak, K.
17 F. Ising Pairing in Superconducting NbSe₂ Atomic Layers. *Nat. Phys.* 2016, 12, 139-143.
- 18 24. Staley, N. E.; Wu, J.; Eklund, P.; Liu, Y.; Li, L. J.; Xu, Z. Electric Field Effect On
19 Superconductivity in Atomically Thin Flakes of NbSe₂. *Phys. Rev. B* 2009, 80.
- 20 25. Frindt, R. F. Superconductivity in Ultrathin Nbse₂ Layers. *Phys. Rev. Lett.* 1972, 28, 299.
- 21 26. Xi, X.; Zhao, L.; Wang, Z.; Berger, H.; Forró, L.; Shan, J.; Mak, K. F. Strongly Enhanced Charge-
22 Density-Wave Order in Monolayer NbSe₂. *Nat. Nanotechnol.* 2015, 10, 765-769.
- 23 27. Naik, I.; Rastogi, A. K. Charge Density Wave and Superconductivity in 2H-and 4H-NbSe₂: A
24 revisit. *Pramana* 2011, 76, 957-963.
- 25 28. Huntley, D. J. Charge-Density Waves and Superconductivity in NbSe₂. *Phys. Rev. Lett.* 1976,
26 36, 490-491.
- 27 29. Xi, X. X.; Zhao, L.; Wang, Z. F.; Berger, H.; Forro, L.; Shan, J.; Mak, K. F. Strongly Enhanced
28 Charge-Density-Wave Order in Monolayer NbSe₂. *Nat. Nanotechnol.* 2015, 10, 765-769.
- 29 30. Huang, Y. H.; Chen, R. S.; Zhang, J. R.; Huang, Y. S. Electronic Transport in NbSe₂ Two-
30 Dimensional Nanostructures: Semiconducting Characteristics and Photoconductivity. *Nanoscale*
31 2015, 7, 18964-18970.
- 32 31. Cao, Y.; Mishchenko, A.; Yu, G. L.; Khestanova, E.; Rooney, A. P.; Prestat, E.; Kretinin, A. V.;
33 Blake, P.; Shalom, M. B.; Woods, C.; Chapman, J.; Balakrishnan, G.; Grigorieva, I. V.; Novoselov, K. S.;
34 Piot, B. A.; Potemski, M.; Watanabe, K.; Taniguchi, T.; Haigh, S. J.; Geim, A. K.; Gorbachev, R. V.
35 Quality Heterostructures from Two-Dimensional Crystals Unstable in Air by Their Assembly in Inert
36 Atmosphere. *Nano Lett.* 2015, 15, pp 4914-4921.
- 37 32. Novoselov, K. S.; Jiang, D.; Schedin, F.; Booth, T. J.; Khotkevich, V. V.; Morozov, S. V.; Geim,
38 A. K. Two-Dimensional Atomic Crystals. *Proc. Natl. Acad. Sci. U.S.A.* 2005, 102, 10451-10453.
- 39 33. Two-Dimensional Nanosheets Produced by Liquid Exfoliation of Layered Materials. *Science*
40 2011, 331, 568.
- 41 34. Light-Induced Oxidation of NbSe₂ Single Crystals. *J. Phys. Chem. Solids* 1971, 32, 2645.
- 42 35. Gao, J.; Li, B.; Tan, J.; Chow, P.; Lu, T.-M.; Koratkar, N. Aging of Transition Metal
43 Dichalcogenide Monolayers. *ACS Nano* 2016, 10, pp 2628-2635.
- 44 36. Arguello, C. J.; Chockalingam, S. P.; Rosenthal, E. P.; Zhao, L.; Gutiérrez, C.; Kang, J. H.; Chung,
45 W. C.; Fernandes, R. M.; Jia, S.; Millis, A. J.; Cava, R. J.; Pasupathy, A. N. Visualizing the Charge
46 Density Wave Transition in 2H-NbSe₂ in Real Space. *Phys. Rev. B* 2014, 89, 235115.
- 47 37. Lovejoy, T. C.; Ramasse, Q. M.; Falke, M.; Kaepfel, A.; Terborg, R.; Zan, R.; Dellby, N.;
48 Krivanek, O. L. Single Atom Identification by Energy Dispersive X-ray Spectroscopy. *Appl. Phys. Lett.*
49 2012, 100, 154101.
- 50 38. D'Alfonso, A. J.; Freitag, B.; Klenov, D.; Allen, L. J. Atomic-Resolution Chemical Mapping Using
51 Energy-Dispersive X-ray Spectroscopy. *Phys. Rev. B* 2010, 81, 100101.
- 52 39. Smith, R. J.; King, P. J.; Lotya, M.; Wirtz, C.; Khan, U.; De, S.; O'Neill, A.; Duesberg, G. S.;
53 Grunlan, J. C.; Moriarty, G.; Chen, J.; Wang, J. Z.; Minett, A. I.; Nicolosi, V.; Coleman, J. N. Large-Scale
54
55
56
57
58
59
60

1
2
3 Exfoliation of Inorganic Layered Compounds in Aqueous Surfactant Solutions. *Adv. Mater.* 2011, 23,
4 3944-3948.

5 40. Ibrahim, M. A.; Huang, W. C.; Lan, T. W.; Boopathi, K. M.; Hsiao, Y. C.; Chen, C. H.;
6 Budiawan, W.; Chen, Y. Y.; Chang, C. S.; Li, L. J.; Tsai, C. H.; Chu, C. W. Controlled Mechanical
7 Cleavage of Bulk Niobium Diselenide to Nanoscaled Sheet, Rod, and Particle Structures for Pt-Free
8 Dye-Sensitized Solar Cells. *J. Mater. Chem. A* 2014, 2, 11382-11390.

9 41. Kretinin, A. V.; Cao, Y.; Tu, J. S.; Yu, G. L.; Jalil, R.; Novoselov, K. S.; Haigh, S. J.; Gholinia, A.;
10 Mishchenko, A.; Lozada, M.; Georgiou, T.; Woods, C. R.; Withers, F.; Blake, P.; Eda, G.; Wirsig, A.;
11 Hucho, C.; Watanabe, K.; Taniguchi, T.; Geim, A. K.; Gorbachev, R. V. Electronic Properties of
12 Graphene Encapsulated with Different Two-Dimensional Atomic Crystals. *Nano Lett.* 2014, 14, 3270-
13 3276.

14 42. Cowley, J. M.; Moodie, A. F. The Scattering of Electrons by Atoms and Crystals .1. A New
15 Theoretical Approach. *Acta Crystallogr.* 1957, 10, 609-619.

16 43. Meyer, J. C.; Geim, A. K.; Katsnelson, M. I.; Novoselov, K. S.; Booth, T. J.; Roth, S. The
17 Structure of Suspended Graphene Sheets. *Nature* 2007, 446, 60-63.

18 44. Hernandez, Y.; Nicolosi, V.; Lotya, M.; Blighe, F. M.; Sun, Z. Y.; De, S.; McGovern, I. T.;
19 Holland, B.; Byrne, M.; Gun'ko, Y. K.; Boland, J. J.; Niraj, P.; Duesberg, G.; Krishnamurthy, S.;
20 Goodhue, R.; Hutchison, J.; Scardaci, V.; Ferrari, A. C.; Coleman, J. N. High-Yield Production of
21 Graphene by Liquid-Phase Exfoliation of Graphite. *Nat. Nanotechnol.* 2008, 3, 563-568.

22 45. Wu, R. J.; Odlyzko, M. L.; Mkhoyan, K. A. Determining the Thickness of Atomically Thin MoS₂
23 and WS₂ in the TEM. *Ultramicroscopy* 2014, 147, 8-20.

24 46. Castellanos-Gomez, A.; Vicarelli, L.; Prada, E.; Island, J. O.; Narasimha-Acharya, K. L.; Blanter,
25 S. I.; Groenendijk, D. J.; Buscema, M.; Steele, G. A.; Alvarez, J. V.; Zandbergen, H. W.; Palacios, J. J.;
26 Zant, H. S. J. v. d. Isolation and Characterization of Few-Layer Black Phosphorus. *2D Mater.* 2014, 1,
27 025001.

28 47. Coleman, J. N.; Lotya, M.; O'Neill, A.; Bergin, S. D.; King, P. J.; Khan, U.; Young, K.; Gaucher,
29 A.; De, S.; Smith, R. J.; Shvets, I. V.; Arora, S. K.; Stanton, G.; Kim, H.-Y.; Lee, K.; Kim, G. T.; Duesberg,
30 G. S.; Hallam, T.; Boland, J. J.; Wang, J. J.; Donegan, J. F.; Grunlan, J. C.; Moriarty, G.; Shmeliov, A.;
31 Nicholls, R. J.; Perkins, J. M.; Grievson, E. M.; Theuwissen, K.; McComb, D. W.; Nellist, P. D.; Nicolosi,
32 V. Two-Dimensional Nanosheets Produced by Liquid Exfoliation of Layered Materials. *Science* 2011,
33 331, 568-571.

34 48. Lin, Y.-C.; Komsa, H.-P.; Yeh, C.-H.; Björkman, T.; Liang, Z.-Y.; Ho, C.-H.; Huang, Y.-S.; Chiu, P.-
35 W.; Krasheninnikov, A. V.; Suenaga, K. Single-Layer ReS₂: Two-Dimensional Semiconductor with
36 Tunable In-Plane Anisotropy. *ACS Nano* 2015, 9, 11249-11257.

37 49. Komsa, H. P.; Krasheninnikov, A. V. Native Defects in Bulk and Monolayer MoS₂ from First
38 Principles. *Phys. Rev. B* 2015, 91, 125304.

39 50. Haigh, S. J.; Gholinia, A.; Jalil, R.; Romani, S.; Britnell, L.; Elias, D. C.; Novoselov, K. S.;
40 Ponomarenko, L. A.; Geim, A. K.; Gorbachev, R. Cross-Sectional Imaging of Individual Layers and
41 Buried Interfaces of Graphene-Based Heterostructures and Superlattices. *Nat. Mater.* 2012, 11, 764-
42 767.

43 51. Zan, R.; Ramasse, Q. M.; Jalil, R.; Georgiou, T.; Bangert, U.; Novoselov, K. S. Control of
44 Radiation Damage in MoS₂ by Graphene Encapsulation. *ACS Nano* 2013, 7, 10167-10174.

45 52. Algara-Siller, G.; Kurasch, S.; Sedighi, M.; Lehtinen, O.; Kaiser, U. The Pristine Atomic
46 Structure of MoS₂ Monolayer Protected from Electron Radiation Damage by Graphene. *Appl. Phys.*
47 *Let.* 2013, 103203107.

48 53. Hýtch, M. J.; Snoeck, E.; Kilaas, R. Quantitative Measurement of Displacement and Strain
49 Fields from HREM Micrographs. *Ultramicroscopy* 1998, 74, 131-146.

50 54. Meyer, J. C.; Eder, F.; Kurasch, S.; Skakalova, V.; Kotakoski, J.; Park, H. J.; Roth, S.; Chuvilin,
51 A.; Eychens, S.; Benner, G.; Krasheninnikov, A. V.; Kaiser, U. Accurate Measurement of Electron
52 Beam Induced Displacement Cross Sections for Single-Layer Graphene. *Phys. Rev. Lett.* 2012, 108,
53 196102.

- 1
2
3 55. Zan, R.; Bangert, U.; Ramasse, Q.; Novoselov, K. S. Interaction of Metals with Suspended
4 Graphene Observed by Transmission Electron Microscopy. *J. Phys. Chem. Lett.* 2012, 3, 953-958.
- 5 56. Berry, V. Impermeability of Graphene and its Applications. *Carbon* 2013, 62, 1-10.
- 6 57. Bunch, J. S.; Verbridge, S. S.; Alden, J. S.; van der Zande, A. M.; Parpia, J. M.; Craighead, H. G.;
7 McEuen, P. L. Impermeable Atomic Membranes from Graphene Sheets. *Nano Lett.* 2008, 8, 2458-
8 2462.
- 9 58. Hu, S.; Lozada-Hidalgo, M.; Wang, F. C.; Mishchenko, A.; Schedin, F.; Nair, R. R.; Hill, E. W.;
10 Boukhvalov, D. W.; Katsnelson, M. I.; Dryfe, R. A. W.; Grigorieva, I. V.; Wu, H. A.; Geim, A. K. Proton
11 Transport Through One-Atom-Thick Crystals. *Nature* 2014, 516, 227-230.
- 12 59. Hanlon, D.; Backes, C.; Doherty, E.; Cucinotta, C. S.; Berner, N. C.; Boland, C.; Lee, K.; Harvey,
13 A.; Lynch, P.; Gholamvand, Z.; Zhang, S.; Wang, K.; Moynihan, G.; Pokle, A.; Ramasse, Q. M.; McEvoy,
14 N.; Blau, W. J.; Wang, J.; Abellan, G.; Hauke, F.; Hirsch, A.; Sanvito, S.; O'Regan, D. D.; Duesberg, G. S.;
15 Nicolosi, V.; Coleman, J. N. Liquid Exfoliation of Solvent-Stabilized Few-Layer Black Phosphorus for
16 Applications Beyond Electronics. *Nat. Commun.* 2015, 6, 8563.
- 17 60. Longchamp, J.-N.; Escher, C.; Fink, H.-W. Ultraclean Freestanding Graphene by Platinum-
18 Metal Catalysis. *J. Vac. Sci. Technol., B: Microelectron. Nanometer Struct. Process., Meas., Phenom.*
19 2013, 31, 020605.
- 20 61. Sarkar, D.; Xie, X.; Kang, J.; Zhang, H.; Liu, W.; Navarrete, J.; Moskovits, M.; Banerjee, K.
21 Functionalization of Transition Metal Dichalcogenides with Metallic Nanoparticles: Implications for
22 Doping and Gas-Sensing. *Nano Lett.* 2015, 15, 2852-2862.
- 23 62. Giovanni, M.; Poh, H. L.; Ambrosi, A.; Zhao, G.; Sofer, Z.; Sanek, F.; Khezri, B.; Webster, R. D.;
24 Pumera, M. Noble Metal (Pd, Ru, Rh, Pt, Au, Ag) Doped Graphene Hybrids for Electrocatalysis.
25 *Nanoscale* 2012, 4, 5002-5008.
- 26 63. Ramasse, Q. M.; Zan, R.; Bangert, U.; Boukhvalov, D. W.; Son, Y.-W.; Novoselov, K. S. Direct
27 Experimental Evidence of Metal-Mediated Etching of Suspended Graphene. *ACS Nano* 2012, 6, 4063-
28 4071.
- 29 64. Krivanek, O.; Chisholm, M.; Nicolosi, V.; Pennycook, T.; Corbin, G.; Dellby, N.; Murfitt, M.;
30 Own, C.; Szilagy, Z.; Oxley, M.; Pantelides, S.; Pennycook, S. Atom-By-Atom Structural and Chemical
31 Analysis by Annular Dark-Field Electron Microscopy. *Nature* 2010, 464, 571-574.
- 32 65. Zan, R.; Bangert, U.; Ramasse, Q.; Novoselov, K. S. Metal-Graphene Interaction Studied *via*
33 Atomic Resolution Scanning Transmission Electron Microscopy. *Nano Lett.* 2011, 11, 1087-1092.
- 34 66. Hu, X.; Wan, N.; Sun, L.; Krasheninnikov, A. V. Semiconductor to Metal to Half-Metal
35 Transition in Pt-Embedded Zigzag Graphene Nanoribbons. *J. Phys. Chem. C*, 2014, 118, 16133-16139.
- 36 67. Rodriguez-Manzo, J. A.; Cretu, O.; Banhart, F. Trapping of Metal Atoms in Vacancies of
37 Carbon Nanotubes and Graphene. *ACS Nano* 2010, 4, 3422-3428.
- 38 68. Stadelmann, P. A. EMS - A Software Package for Electron Diffraction Analysis and HREM
39 Image Simulation In Materials Science. *Ultramicroscopy* 1987, 21, 131-145.
- 40 69. Koch, C. T. Determination Of Core Structure Periodicity And Point Defect Density Along
41 Dislocations. PhD, Arizona State University, 2002.
- 42 70. Kresse, G.; Joubert, D. From Ultrasoft Pseudopotentials to The Projector Augmented-Wave
43 Method. *Phys. Rev. B*. 1999, 59, 1758.
- 44 71. Kresse, G.; Furthmuller, J. Efficiency of Ab-Initio Total Energy Calculations for Metals and
45 Semiconductors Using a Plane-Wave Basis Set. *Comput. Mater. Sci.* 1996, 6, 15.
- 46 72. Perdew, J. P.; Burke, K.; Ernzerhof, M. Generalized Gradient Approximation Made Simple.
47 *Phys. Rev. Lett.* 1996, 77, 3865.
- 48
49
50
51
52
53
54
55
56
57
58
59
60